

CREDIT COURSE OUTLINE

I. COVER PAGE

(1) CHEM 8 (2) ELEMENTARY ORG				ANIC CI	HEMISTRY	(3) 3	
Number			Title		Units		
(4)	Lecture / Lab Hours:			(8)Classification:			
	Total Course Hour	S					
		Total Lec hours:	3.00			Degree applicable:	X
	Total Lab hours: 0		Non-degree applicable:				
	Total Contact hours: 54.00					Basic skills:	
		Lec will generate <u>0</u> hour(s) outside work.		(9)RC	Fulfills AS/AA	A degree requirement: (area)
	Lab will generate _	0 hour(s) outside work.			C	· · · · · · · · · · · · · · · · · · ·	
					General educat		
(5)	Grading Basis:	Grading Scale Only				Area A Natural Sciences	
		Pass/No Pass option	X		Major:		
		Pass/No Pass only			Certificate of:		
(6)	Advisories:				Certificate in:		
	Eligibility for Engl	lish 126		(10)CS		Baccalaureate:	X
	Eligibility for English 125			(11)Repeatable: (A course may be repeated three times)			0
(7)	Pre-requisites(requires C grade or better): CHEM 1A or CHEM 3A			-			
	Corequisites:						
A s		on: ant classes of organic compo n introduction to organic che					

II. COURSE OUTCOMES:

(Specify the learning skills the student demonstrates through completing the course and link critical thinking skills to specific course content and objectives.)

Upon completion of this course, students will be able to:

- I. Draw a structural formula of an organic compound given the systematical name.
- II. Analyze the structural formula of an organic compound, recognize its functional groups and name it properly.
- III. Identify S and R stereoisomers.
- IV. Complete the reactions of simple aliphatic and aromatic molecules, showing the reaction mechanisms.
- V. Analyze simple IR and NMR spectra to determine the structure of an unknown compound.

III. COURSE OBJECTIVES:

(Specify major objectives in terms of the observable knowledge and/or skills to be attained.)

In the process of completing this course, students will:

- I. Learn how to analyze the structural formula and line-bond formula of an organic compound, recognize its functional groups and learn how to name it properly using the IUPAC nomenclature.
- II. Gain an understanding of how to draw structural formulas and line-bond formulas given the systematical name of an organic compound.
- III. Gain a basic understanding of isomers and stereochemistry, recognizing asymmetric carbon atoms that cause chirality.
- IV. Learn how to complete the reactions of simple aliphatic and aromatic molecules and practice writing correct reaction mechanisms.
- V. Learn how to analyze simple IR and NMR spectra and determine the structure of an unknown compound.

IV. COURSE OUTLINE:

Lecture Content:

- A. Covalent bonding and shapes of molecules. Lewis structures and formal charges. Polar covalent and non-polar covalent bonds, intermolecular forces, and their effects on physical properties of organic molecules. Hybridization of molecular orbitals (sp, sp^2 , and sp^3).
- B. Acids and Bases. Lewis and Bronsted definitions.
- C. Alkanes and cycloalkanes. The basics of organic nomenclature.
- D. Chirality and stereo-isomerism. Asymmetric carbon atoms and their effects on stereochemical behavior.
- E. Alkenes and alkynes. Nomenclature of organic molecules containing double and triple bonds. Classification of isomers using the *cis/trans* and *E/Z* notation systems.
- F. Reactions of alkenes. The rule of Markovnikov and its mechanistic background.
- G. Halo-alkanes. Nucleophilic substitution reactions. Replacement of the halogen by nucleophiles such as cyanide, alkoxide, and azide, including the reaction mechanisms.
- H. Alcohols, ethers, and thiols. Nomenclature and synthesis, including reaction mechanisms.
- I. Benzene and its derivatives. Nomenclature and reactivity of substituted benzene rings, including reaction mechanisms.
- J. Aldehydes, ketones, amines and carboxylic acids. Nomenclature and simple reactions, including reaction mechanisms.
- K. Structure determination. Infrared and nuclear magnetic resonance spectroscopy. Analysis of simple IR and NMR spectra using the most modern techniques.
- L. Bio-molecules. An introduction to carbohydrates, lipids, amino acids, proteins and nucleic acids.

V. APPROPRIATE READINGS

Reading assignments may include but are not limited to the following:

- I. Sample Text Title:
 - 1. Recommended Bruice Essentials of Organic Chemistry, ed. 2nd Pearson, Prentice Hall, 2009, ISBN: 0321596951
 - 2. Recommended Bailey and Bailey *Organic Chemistry, A Brief Survey of Concepts and Applications*, ed. 6th Prentice Hall, 2004, ISBN: 0139241191
- II. Other Readings
 - 1. Recommended Luceigh, Organic Chem TV CD-ROM and Lampman Organic Nomenclature CD-ROM
 - 2. Recommended McMurray, Fundamentals of Organic Chemistry, 6th Edition, Thomson, 20027, ISBN 0495012033

 Global or international materials or concepts are appropriately included in this cours
 Multicultural materials and concepts are appropriately included in this course

If either line is checked, write a paragraph indicating specifically how global/international and/or multicultural materials and concepts relate to content outline and/or readings.

VI. METHODS TO MEASURE STUDENT ACHIEVEMENT AND DETERMINE GRADES:

Students in this course will be graded in at least one of the following four categories. Please check those appropriate. A degree applicable course must have a minimum of one response in category A, B, or C.

A. Writing				
	Check either 1 or 2 below			
	1. Substantial writing assignments are required. Check the appropriate boxes below and provide a written description in the space provided.			
	2. Substantial writing assignments are NOT required. If this box is checked leave this section blank. For degree applicable courses you must complete category B and/or C.			
X	a) essay exam(s) X		d) written homework	
	b) term or other paper(s)		e) reading reports	
	c) laboratory report(s)		f) other (specify)	

Required assignments may include but are not limited to the following:

Study questions similar to those in the textbook.

Comprehend journal and newspaper articles and paraphrase ideas.

B. Problem Solving Computational or non-computational problem-solving demonstrations, including:			
X	a) exam(s)		d) laboratory reports
X	b) quizzes		e) field work

X	c) homework problems		1) other (specify):		
	ired assignments may include but are not tions requiring evaluation of data and appli				
C S	kill demonstrations, including:				
C. 5	a) class performance(s)		c) performance exams(s)		
	b) field work		d) other (specify)		
Rogu	ired assignments may include but are not	limita	IL.		
кеци	irea assignments may include but are not	umue	i to the jouowing.		
D. C	Dbjective examinations including:				
	a) multiple choice		d) completion		
	b) true/false	X	e) other (specify):		
	c) matching items				
Description of the control of the co	RSE GRADE DETERMINATION: ription/Explanation: Based on the categories methods fall within the following departicular instructor. The instructor's syllabus not grades must be recorded on the final rost eral methods to measure student achievement final grades.	tmenta nust re- ter.) ent are ments:	I guidelines; however, the final method of flect the criteria by which the student's g used, indicate here the approximate wei	of grading is still at the digrade has been determined	scretion of the I. (A minimum of
	egree applicable courses, the adopted texts in college-level materials.				
Valid	ation Language Level (check where applic	able):		College-Level (YES	Criteria Met NO
Ref Inst	tbook erence materials tructor-prepared materials dio-visual materials			X X X X X	
Indica	used readability formulae (grade level Text is used in a college-level course Used grading provided by publisher Other: (please explain; relate to Skills I		<u>X</u>		
Comp Conte	outation Level (Eligible for MATH 101 levent	el or h	igher where applicable)	<u>X</u>	
Bree Prese Record App pro List of Record Record Record Record Record Record Prese Record Reco	adth of ideas covered clearly meets college entation of content and/or exercises/project quires a variety of problem-solving strategiquires independent thought and study plies transferring knowledge and skills appiblems. If Reading/Educational Materials mmended - Bruice Essentials of Organic Commended - Bailey and Bailey Organic Chec: 0139241191	s: es incl ropriat Themisi	uding inductive and deductive reasoning ely and efficiently to new situations or try, ed. 2nd Pearson, Prentice Hall, 2009,	X X , ISBN: 0321596951	all, 2004,
Comi	ments:				
	This course requires special or addit This course requires special facilitie	s:	ibrary materials (list attached).		
	 Computer lab with programs for cou 	ırse			

	BASIC SKILLS ADVISORIES PAGE The skills listed are those needed for eligibility for English 125, 126, and Math 101. These skills are listed as the outcomes from English 252, 262, and Math 250. In the right hand column, list at least three major basic skills needed at the beginning of the target course and check off the corresponding basic skills listed at the left.				
(eligibility for English 126) (as outcomes for English 262)		1. Students are required to read a college level textbook.			
	X Using phonetic, structural, contextual, and	2. Students are required to use outlining skills and specific textbook study skills.			
	dictionary skills to attack and understand words.	3. Good comprehension of organic reactions requires basic analytical reading skills such as inferring, conclusion etc.			
X Applying word analysis skills to reading in context.					
X Using adequate basic functional vocabulary skills.					
	X Using textbook study skills and outlining skills.				
	X Using a full range of literal comprehension skills and				
	basic analytical skills such as predicting, inferring,				
	concluding, and evaluating.				
		1. Written answers are part of the homework questions.			
	(as outcomes for English 252)				

Writing complete English sentences and avoiding

errors most of the time.

Using the conventions of English writing: capitalization,

punctuation, spelling, etc.

Using verbs correctly in present, past, future, and

present perfect tenses, and using the correct forms of

common irregular verbs.

X Expanding and developing basic sentence structure with

appropriate modification.

Combining sentences using coordination, subordination,

and phrases.

Expressing the writer's ideas in short personal papers

utilizing the writing process in their development.

- 2. On the exams explanations for chemical phenomena need to be written using complete English sentences.
- 3. Reaction mechanisms need to be explained in a concise manner using correct capitalization, spelling and punctuation.

Check the appropriate spaces.

Eligibility for Math 101 is advisory for the target course.

Eligibility for English 126 is advisory for the target course.

Eligibility for English 125 is advisory for the target course.

If the reviewers determine that an advisory or advisories in Basic Skills are all that are necessary for success in the target course, stop here, provide the required signatures, and forward this form to the department chair, the appropriate associate dean, and the curriculum committee.

CONTENT REVIEW CHEM 3A INTRODUCTORY GENERAL CHEMISTRY **CHEM 1A GENERAL CHEMISTRY**

REQUISITES

Subject Prerequisite -- CHEM 1A GENERAL CHEMISTRY

1. Students learn how to work with Lewis structures and geometrical shapes of molecules. 2. Students learn the principles of chemical bonding. They learn the differences between ionic and covalent bonds. 3. Students learn the principles of balancing oxidation and reduction reactions, as well as completing and balancing other reactions

- Learn how to analyze the structural formula and line-bond formula of an organic compound, recognize its functional groups and learn how to name it properly using the IUPAC nomenclature.
- Gain an understanding of how to draw structural formulas and line-bond formulas given the systematical name of an organic compound.
- Learn how to complete the reactions of simple aliphatic and aromatic molecules and practice writing correct reaction mechanisms.

Subject Prerequisite -- CHEM 3A INTRODUCTORY GENERAL CHEMISTRY

1. Students learn how to work with Lewis structures and geometrical shapes of molecules. 2. Students learn the principles of chemical bonding. They learn the difference between ionic and covalent bonds. 3. Students learn the principles of completing and balancing reactions.

- Learn how to analyze the structural formula and line-bond formula of an organic compound, recognize its functional groups and learn how to name it properly using the IUPAC nomenclature.
- Gain an understanding of how to draw structural formulas and line-bond formulas given the systematical name of an organic compound.
- Learn how to complete the reactions of simple aliphatic and aromatic molecules and practice writing correct reaction mechanisms.

ESTABLISHING PREREQUISITES OR COREQUISITES

Every prerequisite or corequisite requires content review plus justification of at least one of the seven kinds below. Prerequisite courses in communication and math outside of their disciplines require justification through statistical evidence. Kinds of justification that may establish a prerequisite are listed below.

Check one of the following that apply. Documentation may be attached.

- The prerequisite/corequisite is required by law or government regulations.
 Explain or cite regulation numbers:
- The health or safety of the students in this course requires the prerequisite.
- Justification: Indicate how this is so.

 The safety or equipment operation skills learned in the prerequisite course are required for the successful or safe completion of this course.

Justification: Indicate how this is so.

- The prerequisite is required in order for the course to be accepted for transfer to the UC or CSU systems.
 Justification: Indicate how this is so.
- Significant statistical evidence indicates that the absence of the prerequisite course is related to unsatisfactory performance in the target course.

Justification: Cite the statistical evidence from the research.

- 6. X The prerequisite course is part of a sequence of courses within or across a discipline.
- 7. ____ Three CSU/UC campuses require an equivalent prerequisite or corequisite for a course equivalent to the target course: